PHYSICAL SCIENCES

SCOPE FOR EXAMINATIONS 2021: **GRADE 1**0

MARCH CONTROLLED TEST (100 marks)			
PAPER	TOPICS	MARK	DURATION
ONE PAPER	Electricity & Magnetism: ✓ Electrostatics	25	
ONLY	Matter and Materials: ✓ Matter & Classification ✓ States of Matter and the KMT ✓ The Atom ✓ The Periodic Table ✓ Chemical Bonding	75	2 hours
	JUNE CONTROLLED TEST (100 marks)		
	NB: June Controlled Test will assess second term wo	rk only.	1
ONE PAPER ONLY	Electricity & Magnetism: ✓ Electric Circuits Mechanics: ✓ Vectors and Scalars ✓ Motion in 1-dimension ✓ Instantaneous Speed & Velocity ✓ Equations of Motion ✓ Graphs of Motion ✓ Energy & Conservation of Mechanical Energy	100	2 hours

SEPTEMBER CONTROLLED TEST (100 marks)			
PAPER	TOPICS	MARKS	DURATION
ONE PAPER ONLY	Waves, Sound & Light Waves	±40	2 hours
	 ✓ Transverse pulses on a string / spring ✓ Transverse Waves. ✓ Longitudinal Waves. 		
	Sound ✓ Frequency and Pitch ✓ Amplitude and Loudness ✓ Ultrasound		
	Chemical Change: ✓ Physical and Chemical change ✓ Representing Chemical change ✓ Quantitative aspects of Chemical change	±60	

FINAL EXAMINATION: GRADE 10

FORMAT OF QUESTION PAPERS

Paper 1: Physics	Paper 2: Chemistry
2 hours	2 hours
SECTION A:	SECTION A:
Multiple-choice questions	Multiple-choice questions
SECTION B:	SECTION B:
Longer questions assessing all themes	Longer questions assessing all themes
Total: 100 marks	Total: 100 marks

MARK ALLOCATION PER KNOWLEDGE AREA: FINAL EXAMINATIONS GR 10

PAPER 1

Knowledge Area	Theme	Marks
	Vectors & Scalars	
Mechanics	Motion in one dimension	
(± 45%)	Instantaneous Speed, Velocity & Equations of Motion	45
	Energy & Conservation of Mechanical Energy	
Electricity and magnetism	Electrostatics	25
(± 25%)	Electrical Circuits	25
	Transverse pulses on a string or spring	
	Transverse and Longitudinal Waves	20
Waves, Sound & Light	Sound	30
(± 30%)	Electromagnetic Radiation	
	TOTAL	100

PAPER 2

Knowledge Area	Theme	Marks
	Matter & Classification; Kinetic Molecular Theory Particles Substances are made of	
Matter and materials	The Atom: Basic building blocks of Matter	50
(±50 %)	Periodic Table	
	Chemical Bonding	
Chemical change	Physical & Chemical Change	
(± 50%)	Representing Chemical Change	50
	Quantitative aspects of Chemical Change	
	TOTAL	100

COGNITIVE LEVELS

Cognitive level description	Weighting %	
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Remembering	15	15
Understanding	30	40
Applying and Analysing	45	35
Evaluating	10	10