#### 2021 National ATP: Grade 11 Term 1: TECHNICAL SCIENCES

# basic education Department: Basic Education REPUBLIC OF SOUTH AFRICA

#### **Important notes**

- 1. The content of the tables are CAPS aligned however the CAPS document must be used at all times for further details.
- 2. The formal assessment will consist of:
  - 2.1 Term 1 Control test 1 (30% of SBA) & Formal experiment 1/PAT1 (40% of PAT)
  - 2.2 Term 2 Control test 2 (40% of SBA) & Formal experiment 2/PAT2 (30% of PAT)
  - 2.3 Term 3 Control test 3 (30% of SBA) & Formal experiment 3/PAT 3 (30% of PAT)
  - 2.4 Term 4 End of the year examination (2 Papers)

TERM 1 (45 days)	Week 1 27 - 29 Jan (3 days)  • MECHANICS: Revision of grade 10 content (3hrs)	Week 2 01 - 05 Feb (5 days)  MECHANICS: Revision of grade 10 content (4hrs)	Week 3 08 –12 Feb (5 days)  MECHANICS: Introduction to Mechanics Sign conventions (4hrs)	Week 4 15 - 19 Feb (5 days)  MECHANICS: Graphs (4hrs)	Week 5 22 - 26 Feb (5 days)  MECHANICS:  • Theorem of Pythagoras  • Co-linear vectors • Co-planar vectors (4hrs)	Week 6 01 - 05 March (5 days)  MECHANICS: Resultant of forces in two dimensions. Head-to-tail method Theorem of Pythagoras (4hrs)	Week 7 08 - 12 March (5 days)  MECHANICS: Resultant of forces in two dimensions. Parallelogram of Forces (4hrs)	Week 8 15 - 19 March (5 days)  MECHANICS: Resolution of forces into components (4hrs)	Week 9 23 - 26 March (4 days)  MECHANICS: Frictional forces Static and kinetic frictional force (4hrs)	Week 10 29 - 31 March (3 days)  Consolidation and revision.  Control test 1 (1hr)
Topics /Concepts, Skills and Values	Vectors and scalars (Vectors, scalars, graphical representation of vectors)      Motion in one dimension: (position, displacement, distance, speed, velocity, acceleration)      Introduction of Force (Definition of force, contact force, non-contact force),	Kinds of forces (Tension, normal force, force of gravity, frictional Force)     Force diagram and free body diagram,     Resultant and Equilibrant     Equilibrium of forces in one dimension     Energy (gravitational potential energy, kinetic energy, mechanical energy)	Use the Cartesian coordinates system to indicate the directions (+ve X and +ve Y as positive).  Use compass directions to indicate the directions.  Express the direction using bearing by measuring on the north line in the clockwise direction to the vector.  Use the above methods to determine the directions of vectors.	Demonstrate the direct proportion graphs in the context of technology.      Demonstrate the indirect proportion graphs in the context of technology.	<ul> <li>Determine the resultant of two vectors acting perpendicular to each other using the theorem of Pythagoras: F<sub>R</sub><sup>2</sup> = F<sub>1</sub><sup>2</sup> + F<sub>2</sub><sup>2</sup></li> <li>Use the theorem of Pythagoras to calculate the resultant of forces, in the context of technology.</li> <li>Define co-linear vectors as vectors that have the same line of action.</li> <li>Define co-planar vectors as vectors that are in the same plane.</li> <li>Draw the resultant of two co-linear vectors.</li> </ul>	Use the head-to-tail method to determine the resultant of two vectors at right angles to each other.  Use the theorem of Pythagoras to determine the resultant of forces acting at right angles to each other.	The Parallelogram law of forces states that if two forces acting at a point can be represented by the adjacent sides of a parallelogram both in magnitude and direction, then the diagonal from the point gives the resultant of the two forces.  Use the parallelogram law to determine the resultant of two forces acting at an angle to each other.  Using scale drawing (Do not do calculations involving the resultant.)	Resolution of forces:  • Given a force F acting at an angle to the horizontal axis, resolve the force into its parallel and perpendicular components. (use scale drawings)  • Given a force F acting at an angle to the horizontal axis, resolve the force into its parallel and perpendicular components. (use calculations)	<ul> <li>Frictional forces:</li> <li>Define frictional force as the force that opposes the motion of an object.</li> <li>The static (limiting) frictional force acts between the two surfaces when the object is stationary. It is given by f<sub>S</sub> = μ<sub>S</sub>F<sub>N</sub> Use the above equation to solve problems involving frictional forces. (No inclined plane problems)  The kinetic (dynamic) frictional force acts between the two surfaces when the object is moving. It is given by f<sub>k</sub> = μ<sub>k</sub>F<sub>N</sub> Use the above equation to solve problems involving frictional forces. (No inclined plane problems)</li> <li>Experiment 2 (Informal) a) Determine the relation between the force of limiting friction and the normal force.</li> <li>b) Determine the coefficient of friction between a block and horizontal surface.</li> </ul>	Control test 1 (1hr) Signs and conversions Graphs Theorem of Pythagoras an its application Co-linear vectors Co-planar vectors Resultant of forces in two dimensions Resolution of forces into components Frictional forces
Requisite pre- knowledge			Graphical representation of vectors.	Basic skills on drawing and interpreting graphs.	Graphical representation of vectors. Working with formulae.	Graphical representation of vectors.	Graphical representation of vectors.	Graphical representation of vectors.	Graphical representation of vectors. Kinds of forces. Working with formulae	

1



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	TERM 1 (45 days)	Week 1 27 - 29 Jan (3 days)	Week 2 01 - 05 Feb (5 days)	Week 3 08 –12 Feb (5 days)	Week 4 15 - 19 Feb (5 days)	Week 5 22 - 26 Feb (5 days)	Week 6 01 - 05 March (5 days)	Week 7 08 - 12 March (5 days)	Week 8 15 - 19 March (5 days)	Week 9 23 - 26 March (4 days)	Week 10 29 - 31 March (3 days)
Resources (other than textbook) to enhance learning				Question bank such as previous papers or study guides Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos				
ment	Informal Assessment: Remediation	Homework	Homework	Homework Informal test	Homework	Homework	Homework Informal test			Homework	Homework Experiment 2
Assess	SBA (Formal)	None	None	None	None		None	Formal Experiment 1 (PAT 1)	None	None	Control test 1 (1hr)



### 2020 National ATP: Grade 11 – Term 2: TECHNICAL SCIENCES

	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	Week 7	Week 8	Weeks 9	Weeks 10	Weeks 11
TERM 2	13 – 16 April	19 – 23 April	28 – 30 April	3 – 7 May	10 – 14 May	17 – 21 May	24 – 28 May	31 May – 4 June	7 – 11 June	14 – 18 June	21 – 25 June
(51 days)	(4 days)	(5 days)	(3 days)	(5 days)	(5 days)	(5 days)	(5 days)	(5 days)	(5 days)	(4 days)	(5 days)
CAPS Topics	MAGNETISM AND ELECTRICITY: Magnet and the magnetic field (4hrs)	MAGNETISM AND ELECTRICITY: The earth's magnetic field. (4hrs)	WAVES AND SOUND: Pulses (4hrs)	WAVES AND SOUND: • Pulses (2hrs) • Waves (2hrs)	WAVES AND SOUND: • Waves (1hr) • Wave terminology (3hrs)	WAVES AND SOUND: (4hrs)	WAVES AND SOUND: Sound waves (4hrs)	WAVES AND SOUND: (4hrs)	WAVES AND SOUND: (4hrs)	MATTER AND MATERIALS Electrical conductors, semiconductors and insulators. (Revision)	Consolidation, Revision and control test 2
Topics /Concepts, Skills and Values	<ul> <li>Describe a magnet as an object that has a pair of opposite poles, called north and south. Even if the object is cut into tiny pieces, each piece will still have both a N and a S pole.</li> <li>Define the magnetic field as the region in space where another</li> <li>Magnet or ferromagnetic material will experience a force.</li> <li>Like magnetic poles repel each other and opposite poles attract each other.</li> <li>Use a compass to determine the direction of the magnetic field.</li> <li>Sketch the magnet.</li> <li>Predict the behaviour of magnets when they are brought close together.</li> <li>Discuss the properties of magnetic field lines.</li> </ul>	<ul> <li>Compare the magnetic field of the earth to the magnetic field of a bar magnet.</li> <li>Explain the difference between the geographical North pole and the magnetic North pole of the earth.</li> <li>Give examples of phenomena that are affected by earth's magnetic field e.g. Aurora Borealis (Northern Lights) &amp; magnetic storms.</li> <li>Discuss qualitatively how the earth's magnetic field provides protection from solar winds.</li> </ul>	Define a pulse as a single disturbance in a medium.      Define a transverse pulse as a pulse in which the particles of the medium vibrate at right angles to the direction of propagation of the pulse.      Define a longitudinal pulse as a pulse in which the particles of the medium vibrate parallel to the direction of propagation of the pulse.	Pulses Experiment 6 (Spend 2hr) Observe the motion of a single pulse  Waves (2 hr) • Define a wave as a succession of pulses. • Define a transverse wave as a wave in which the particles of the medium vibrate at right angles to the direction of propagation of the wave. • Draw the transverse wave.	longitudinal wave as a wave in which the particles of the medium vibrate parallel to the direction of propagation of the wave.  • Draw the longitudinal wave.  • Define amplitude as the maximum displacement of a particle from its rest (equilibrium) position.  • Define a crest as the uppermost point on a transverse wave.  • Define a trough as the lowermost point on a transverse wave.  • Define points in phase as any two points that are in the same state of vibration.  • Define wavelength (as the distance between two successive points in phase. SI unit: m  • Draw and label transverse and longitudinal waves.  • Define the period (T) as the time taken to complete one wave.  SI unit: s  • Define frequency (f) as the number of waves per second. SI unit: hertz (Hz) Note: 1 Hz = 1 s <sup>-1</sup>	Relationship between period and frequency:  • $T = \frac{1}{f}$ • Use the above equation to solve problems involving period and frequency in the content of technology.  Wave speed:  • Define wave speed as the distance travelled by the wave in one second. $V = \frac{distance\ travelled\ time\ taken}{T}$ or $V = \frac{\lambda}{T}$ or $V = f\lambda$ • Use the above equations to solve problems involving speed, wavelength and frequency, distance, time, in the content of technology.	Sound waves:  Sound waves are longitudinal waves.  Investigate the speed of sound waves in different mediums (gas, liquid or solid).  Define the reflection of sound waves as the bouncing back of the wave from a surface.  Define an echo as the reflection of a sound wave.	<ul> <li>Define pitch as a measure of how high or low a note is.</li> <li>Frequency of sound determines its pitch. The higher the frequency, the higher the pitch.</li> <li>Loudness is determined by the amplitude of the sound.</li> <li>The higher the amplitude, the louder sound.</li> <li>Use wave patterns to demonstrate pitch and loudness.</li> <li>Infrasound: frequencies less than 20 Hz.</li> <li>Audible sound: frequencies from 20 Hz to 20 000 Hz.</li> <li>Ultrasound: frequencies greater than 20 000Hz.</li> <li>Application of infrasound and ultrasound related to technology.</li> </ul>	Formal Experiments (PAT)	Give examples of electrical conductors, semiconductors and insulators.  Identify the substances and the appliances or objects, which are in common daily use in homes and offices, that are specifically chosen because of their electrical properties (conductors, insulators and semiconductors  .	Control test 2 (1hr)  Magnet  The magnetic field  Poles of permanent magnet  Direction of magnetic field  Magnetic field of a bar magnet  Force of a magnet  Properties of magnetic field lines  Earth's Magnetic Field  Pulses  Waves  Wave terminology  Sound waves



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	TERM 2 (51 days)	Week 1 13 – 16 April (4 days)	Week 2 19 – 23 April (5 days)	Week 3 28 – 30 April (3 days)	Week 4 3 – 7 May (5 days)	Week 5 10 – 14 May (5 days)	Week 6 17 – 21 May (5 days)	Week 7 24 – 28 May (5 days)	Week 8 31 May – 4 June (5 days)	Weeks 9 7 – 11 June (5 days)	Weeks 10 14 – 18 June (4 days)	Weeks 11 21 – 25 June (5 days)
	quisite pre- wledge	Magnetic, non- magnetic and ferromagnetic material	Magnetic, non- magnetic and ferromagnetic material					Definitions of frequency and amplitude.	Definitions of frequency and amplitude.	Definitions of frequency and amplitude.		
tha	sources (other n textbook) to ance learning	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides Simulations Videos	Question bank such as previous papers or study guides Simulations Videos	Question bank such as previous papers or study guides	Question bank such as previous papers or study guides Simulations Videos	Question bank such as previous papers or study guides Simulations Videos	Question bank such as previous papers or study guides Simulations Videos		
Assessment	Informal Assessment: Remediation	Homework  Experiment 3 (Informal)  Determine the north pole of the earth using a bar magnet	Homework  Experiment 4 (Informal)  a) Determine whether a material is a magnetic material or a magnet. b) Determine the polarity of the magnets.  Experiment 5 (Informal) Mapping of magnetic field.	Corrections of March control test Homework	Homework  Experiment 6 (Informal):  Observe the motion of a single pulse travelling along a long, soft spring or a heavy rope.	Homework Informal test	Homework	Homework	Homework Informal test		Homework  Informal Experiment (Simulation, video or demonstration)  Determine the electrical conductivity of different material.	
	SBA (Formal)			None	None	None	None	• None	• None	Formal Experiment (PAT 2)		Control test 2

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#### 2021 National ATP: Grade 11 - Term 3: TECHNICAL SCIENCES

2021 National ATP: Grade 11 – Term 3: TECHNICAL SCIENCES											
TERM 3	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	Week 7	Week 8	Week 9	Week 10	Weeks 10 -
(52 days)	13 – 16 July	19 – 23 July	26 – 30 July	2 – 6 Aug	10 – 13 Aug	16 – 20 Aug	23 – 27 Aug	30 Aug-3 Sept	6 – 10 Sept	13 – 17 Sept	20 – 23 Sept
(32 days)	(4 days)	(5 days)	(5 days)	(5 days)	(4 days)	(5 days)	(5 days)	(5 days)	(5 days)	(5 days)	(4 days)
	ELECTRICITY &	ELECTRICITY	ELECTRICITY	ELECTRICITY AND	ELECTRICITY AND	ELECTRICITY AND	ELECTRICITY AND	ELECTRICITY AND	HEAT AND	HEAT AND	HEAT AND
	MAGNETISM Electrostatics	AND MAGNETISM Electrostatics	AND MAGNETISM Electrostatics	MAGNETISM Electric circuits	MAGNETISM Electric circuits	MAGNETISM Electric circuits	MAGNETISM Electric circuits	MAGNETISM Electric circuits	THERMODYNAMICS	THERMODYNAMICS	THERMODYNAMICS
OADO Tarriar	Revision of Grade 10	(4hrs)	(4hrs)	Revision of Grade 10	(4hrs)	(4hrs)	(4hrs)	(4hrs)	(4hrs)	(4hrs)	(2hrs)
CAPS Topics	content (1hr)	(41113)	(41113)	content	(41113)	(41113)	(41113)	(41113)			Control Test 3 (1hr)
	Coulomb's Law										
	(2hr)										
	Two kinds of	Electric field	Electric field	Components of electric	Ohm's Law	Ohmic and non-	Circuit	Emf	Heat: Specific heat	Experiment 13	First Law of
	charge (1hr)	Define the	lines	circuit:	Ohm's law states that	Ohmic conductors:	calculations	<ul> <li>Emf is defined</li> </ul>	capacity	Determine the heat	Thermodynamics
	<ul><li>Explain that all</li></ul>	electric field	Draw electric	Draw the components	the current in a	Any conductor that	<ul> <li>Use series and</li> </ul>	as the	Define the	capacity of a solid.	The first law of
	materials contain	as a region	field lines:	of a circuit using	conductor is directly	obeys Ohm's law is	parallel	potential	specific capacity	(Materials:	thermodynamics
	positive charges	of space in	a) Around a positive	appropriate circuit symbols.	proportional to the potential difference	called an Ohmic conductor.	resistors in	difference	(c) of a substance	Calorimeter,	states that if heat
	(protons) and negative charges	which an	charge	Give the meanings of	across it, at constant	Give examples of	combination	across a cell	as the amount of	thermometer,	energy ΔQ is given
	(electrons).	electric	b) Around a	all symbols used.	temperature.	Ohmic conductors.	with Ohm's law.	when the	heat required to		to a system it is
	• Explain that an	charge	negative	Current:	V= IR	A conductor that		circuit is open	increase the	balance, lead or	used in two ways:
	object which has	experiences	charge	Define current, I.	Use the above	does not obey		<ul> <li>Define Internal</li> </ul>	temperature of 1	sand, water etc.)	(i) In increasing the
	an equal number	a force.	c) Between a	The unit for current is	equation to do	Ohm's law is called		resistance as	kg of the	Thermodynamics	internal energy of
	of electrons and	_	positive and	Ampere (A),  • Calculate the current	calculations (include	non-Ohmic		the resistance	substance by	• In	the system (∆U)
	protons is neutral	$E = \frac{F}{a}$	a positive charge	using the equation	graphical calculations).	conductor. Give examples of non-		inside the cell	1°C or 1K.	thermodynamics	(ii) In doing
	(no net charge).	Q	d) Between a		calculations).	Ohmic conductors.		when current	SI unit: Jkg <sup>-1</sup> K <sup>-1</sup>	we deal with the	work
	<ul> <li>Explain that</li> </ul>		negative and	$I = \frac{Q}{}$	Experiment 10 -	Omino conductore.		flows through	Heat capacity	processes	against
	positively charged	Use the	a negative	Δt	2 hrs	Experiment 11 –		it.	• Define the heat	involving heat,	external
	objects are	above	charge	Direction of	Determine the	2 hrs		(No calculations	capacity (C) of a	work and energy.	pressure.
	electron deficient	equation to	e) Between a		resistance of an	Obtain current		needed)	substance as the	Define	(ΔW).
	and negatively	calculate the	positive and	conventional current in	unknown resistor.	and voltage data			amount of heat	thermodynamic	• $\Delta Q = \Delta U + \Delta W$
	charged objects	force,	a negative charge.	circuit		for a piece of copper wire and		Experiment 12 –	required to increase	system as a	Lies the above
	have an excess of	charge and	charge.	Define potential		semi-conductor		Determine the internal resistance	the temperature of	portion of	Use the above
	<mark>electrons.</mark>	electric field.		difference, Emf		and determine		of a battery.	the whole	matter. E.g.	equation to calculate the
	Coulomb's Law	• The	Electric field	Give the difference		which one obeys		or a ballery.	1	Gas enclosed	
	(2hrs)	direction of	between	between emf and		Ohm's law.			substance by 1°C	inside a	internal energy, work done and the
	Coulomb's Law	the electric	parallel plates.	potential difference.					or 1K	cylinder, fitted	amount of heat
Topics /	states that the	field at a	$E = \frac{V}{I}$	Emf and pd are measured in volts					SI unit: JK <sup>-1</sup>	with a piston.	supplied.
Concepts,	force of attraction	point is the	• Do d	(V).					• C = cm	Define the	Define working
Skills and Values	or repulsion	direction	calculations by	<ul> <li>Do calculations using</li> </ul>					where m is the	surrounding as	substance as the
	between two	that a	using the	the above equations.					mass of a		substance that
	point charges is directly	positive test	above	Resistance					substance.	anything outside the	absorbs heat from
	proportional to the	charge	equation.	Define resistance and					<ul> <li>Use the above</li> </ul>	system which	the source. e.g.
	product of their	(+1C) would	Discuss the	give a microscopic					equation to do	has some	Air in petrol and
	charges and	move if	relationship between E, V	description of resistance					calculations.	bearing on the	diesel engines.
	inversely	placed at that point.	and d.	in terms of electrons					Discuss practical	behaviour of the	<ul> <li>Define heat engine</li> </ul>
	proportional to the square of the	liiat poiiit.	Draw electric	moving through a conductor and colliding					application of heat	system.	as a device which
	distance between		lines between	with the particles of					capacity in	<ul> <li>Define an</li> </ul>	converts heat
	the two charges.		two parallel	which the conductor					technology.	open system	energy into
			plates.	(metal) is made and					Law of conservation of	as a system	mechanical work.
	$F = \frac{kQ_1Q_2}{r^2}$		Discuss application of	thereby transferring					heat	which can	Efficiency of heat
	$r^2$		electrostatics	kinetic energy.						exchange	engine
			related to	State and explain factors     that affect the registered					• Law of	matter and	It absorbs heat
	Use the above		technology.	that affect the resistance of a substance					conservation of	energy with	from a <b>hot body</b>
	equation to calculate			Resistors in Series					heat states that	the	(source), converts
	the force and			$R_T = R_1 + R_2 + R_3$					the amount of heat lost equals	surroundings.	a part of it into
	charge.			$I_T = I_1 = I_2 = I_3$					the amount of	Dofine a	work and rejects
				$V_T = V_1 + V_2 + V_3$					heat gained,	Define a	the rest to a <b>cold</b>
				Resistors in parallel					when no heat is	closed system	body (sink).
				1 1 1 1					lost.	as a system which can	
				$\frac{1}{R_P} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3}$					• Amount of heat	exchange	
									lost or gained is	energy only,	
				Alternatively, when we have two resistors in					given by:	not matter, with	
	<u> </u>	<u> </u>	1	have two resistors in		L	<u> </u>	<u> </u>	giveir by.	Hot matter, with	1



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			parallel we can use the formula. $R_p = \frac{R_1 \times R_2}{R_1 + R_2}$ • $\forall \tau = \forall = \forall$					Q = mc\Deltat SI unit of specific heat capacity: Jkg\(^1\).  • Do calculations using the above equation.  Experiment 13 • Determine the heat capacity of a solid. (Materials: Calorimeter, thermometer, balance, lead or sand, water etc.)	the surroundings.  Define an isolated system as a system which is not influenced by its surroundings. (No exchange of heat or energy with the surroundings).  The thermal state of a system is defined by its temperature (T), pressure (P) and volume (V). These quantities are called thermodynamic variables.  Define internal energy of a thermodynamic system as the sum of the kinetic and potential energies of all the molecules of the system.	Second law of thermodynamics  Efficiency = \frac{w}{Q_1} (No calculation on efficiency of a heat engine)  It is impossible to get a continuous supply of work from a body by cooling it to a temperature lower than the lowest of its surroundings.  It is the reverse of a heat engine.  Refrigerators:  The working substance (coolant e.g. liquid ammonia, Freon etc.) absorbs heat from a cold body (freezer), with the help of an external agency (compressor) and rejects it to the hot body (atmosphere).  Control Test 3 (1hr)  Coulombs law  Electric field lines  Application of electrostatics  Electric circuits  Heat and Thermodynamics
Requisite pre- knowledge	Two kinds of charge and charge conservation.	Two kinds of charge	Two kinds of charge	Components of a circuit, current, potential difference, resistance, resistors in series, resistors in parallel.	Components of a circuit, current, potential difference, resistance, resistors in series, resistors in parallel.	Components of a circuit, current, potential difference, resistance, resistors in series, resistors in parallel.	Components of a circuit, current, potential difference, resistance, resistors in series, resistors in parallel.			
Resources (other than textbook) to enhance learning	Question bank such as previous papers or study guides Videos	Question bank such as previous papers or study guides Videos	Question bank such as previous papers or study guides Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides	such as previous papers or study guides Practical apparatus Simulations Videos	Question bank such as previous papers or study guides



Assessment	Informal Assessment: Remediation	Homework	Informal test	Homework Informal test	Homework	Experiment 11 Obtain current and voltage data for a piece of copper wire and semi-conductor and determine which one obeys Ohm's law	Homework	Homework	Homework	Homework  Experiment 13  Determine the head capacity of a solid. (Materials: Calorimeter, thermometer, balance, lead or sand, water etc.)	
	SBA (Formal)	None	None	None	None	None	Formal Experiment (PAT 3).	None	None	None	Control Test 3 (1hr)



### 2020 National ATP: Grade 11 - Term 4: TECHNICAL SCIENCES

TE	ERM 4	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	Weeks 7 - 10
	' days)	5 – 8 Oct	11 – 15 Oct	18 - 22 Oct	25 – 29 Oct	1 – 5 Nov	8 – 12 Nov	15 Nov – 8 Dec
(47	udysj	(4 days)	(5 days)	(5 days)	(5 days)	(5 days)	(5 days)	(18 days)
		MATTER AND MATERIALS:	MATTER AND MATERIALS:	CHEMICAL CHANGE	CHEMICAL CHANGE	CHEMICAL CHANGE	CHEMICAL CHANGE	Consolidation, Revision and
CAPS Topics	6	Classification of matter	Classification of matter	Oxidation and Reduction	Oxidation and Reduction	Oxidation and Reduction	Oxidation and Reduction	End of the year Examination
		Revision of Grade 10 content	Revision of Grade 10 content	(4 hr)	(4 hr)	(4 hr)	(4 hr)	
Topics /Concepts, Skills and Values		<ul> <li>Classification of matter:</li> <li>Define a pure substance as a single type of material (elements or compounds).</li> <li>Define an element as the simplest type of a pure substance.</li> <li>Define a compound as a substance made up of two or more elements in the exact ratio.</li> <li>Classify substances as pure, compounds or elements.</li> </ul>	Naming of compounds  Name compounds using the names of the elements from which they are made.  Define the terms cation and anion. Identify cations and anions. List the common compound anion, only sulphate, carbonate, sulphite, hydroxide  Molecular fomulae Use cations and anions to write formulae  Balancing of equations Represent reactions in equations and balancing equations	Oxidation is defined as the loss of electrons.     Give examples of oxidation	Reduction is defined as the gain of electrons. Give examples of reduction	<ul> <li>An oxidizing agent is defined as a substance that undergoes reduction.</li> <li>A reducing agent is defined as a substance that undergoes oxidation.</li> <li>Rules for assigning oxidation numbers.</li> <li>Assign oxidation numbers in various molecules.</li> <li>Electrolysis is the decomposition of a substance when an electric current is passed through it.</li> <li>Cathode is the electrode where reduction takes place.</li> <li>Anode is the electrode where oxidation takes place.</li> </ul>	Experiment 15  • Electrolysis of a salt solution.  (Materials: Carbon electrodes, beaker, copper chloride, water, power source, connecting wires, switch, etc.)	All content, concepts and skills as prescribed in the CAPS for terms 1- 4 except  • Superposition of waves  • Paper 1 (150 marks)  • Mechanics (48)  • Electricity and Magnetism (54)  • Waves, sound and light (48)  • Paper 2 (75 marks)  • Chemical Change (38)  • Heat and thermodynamics (37)
Requisite pre	e-knowledge			Structure of the atom (Atomic Number, mass number, The Periodic Table, electron configuration)	Structure of the atom (Atomic Number, mass number, The Periodic Table, electron configuration)	Structure of the atom (Atomic Number, mass number, The Periodic Table, electron configuration)		
Resources (other than textbook) to enhance learning				Question bank such as previous papers or study guides Videos	Question bank such as previous papers or study guides Videos	Question bank such as previous papers or study guides Videos	Question bank such as previous papers or study guides Practical apparatus Simulations Videos	
sessment	Informal Assessment: Remediation			Homework	Homework	Informal test	Experiment 15     Electrolysis of a salt solution.	
Asse	SBA (Formal)			None	None	None	None	End of the year Examination